

## Tutorial sheet 14

### 28. Bulk viscosity

In a hydrodynamic model, the (conserved) stress-energy tensor for a fluid obeying the Navier–Stokes equation can be written as

$$T_{ij}(t, \vec{r}) = \mathcal{P}(t, \vec{r})\delta_{ij} + \pi_{ij}^{\text{shear}}(t, \vec{r}) + \Pi(t, \vec{r})\delta_{ij},$$

where  $\mathcal{P}$  is the equilibrium pressure,  $\pi_{ij}^{\text{shear}}$  the traceless shear stress tensor—which depends on the first derivatives of the flow velocity  $\vec{v}(t, \vec{r})$ —, and  $\Pi$  the bulk pressure. The latter is of the form  $\Pi = \zeta \vec{\nabla} \cdot \vec{v}$ , with  $\zeta$  the bulk viscosity. Thus,  $\zeta$  measures the size of the deviation [in compressible ( $\vec{\nabla} \cdot \vec{v} \neq 0$ ) flows] of the trace  $\sum_i T_{ii}$  of the stress-energy tensor from the equilibrium value  $3\mathcal{P}$ .

Microscopically, the stress-energy tensor for a gas of non-interacting particles is related to the phase space density  $\bar{f}(t, \vec{r}, \vec{p})$  through

$$T_{ij}(t, \vec{r}) \equiv \int p_i p_j \bar{f}(t, \vec{r}, \vec{p}) \frac{d^3 \vec{p}}{(2\pi\hbar)^3 E_{\vec{p}}},$$

where  $E_{\vec{p}}$  denotes the energy of a particle with momentum  $\vec{p}$ , both quantities being measured in the local rest frame of the fluid — i.e. the momentum-space average of  $\vec{p}$ , weighted with  $\bar{f}$ , vanishes.

Similarly, the energy density  $e$ —which is related to the equilibrium pressure with the help of the equation of state of the fluid—is given by

$$e(t, \vec{r}) \equiv \int E_{\vec{p}} \bar{f}(t, \vec{r}, \vec{p}) \frac{d^3 \vec{p}}{(2\pi\hbar)^3} = \int E_{\vec{p}}^2 \bar{f}(t, \vec{r}, \vec{p}) \frac{d^3 \vec{p}}{(2\pi\hbar)^3 E_{\vec{p}}},$$

where the second identity was included to emphasize the analogy with the definition of  $T_{ij}$ .

#### i. Gas of massless particles

For a noninteracting gas of massless particles, the equation of state reads  $e = 3\mathcal{P}$ .<sup>1</sup> Using the known relation for  $E_{\vec{p}}$  as function of momentum, check that, *irrespective of the expression for  $\bar{f}$* , the trace of the stress-energy tensor simply equals  $e$ . What does this mean for the bulk viscosity  $\zeta$ ?

#### ii. Nonrelativistic ideal gas

The internal energy  $U$  and pressure  $\mathcal{P}$  of a nonrelativistic ideal gas of particles with mass  $m$  occupying a volume  $\mathcal{V}$  obey the relation  $U = \frac{3}{2}\mathcal{P}\mathcal{V}$ .<sup>1</sup> Using the nonrelativistic expression for  $E_{\vec{p}}$  and appropriate approximations, compute the trace of the stress-energy tensor and express it as a (simple) function of the energy density  $e$  and the particle number density  $n$  defined as the integral over momenta of the phase space density. What does this again give for the bulk viscosity  $\zeta$ ?

*Hint:* You should recover the expression of the stress-energy tensor given in the lecture, where it is denoted by  $P_{ij}$ ... Remember also what is included in the internal energy  $U$ , and what is not.

**Remark:** Since no knowledge of the phase space density  $\bar{f}$  was actually needed in the above derivations, the systems need not be equilibrated for the result(s) to hold.

### 29. Validity of the hydrodynamic regime

Consider a Boltzmann gas with typical density  $n$ , contained in a volume  $L^3$ . Let  $\ell_{\text{mfp}}$  denote the mean free path of the particles in the gas. A (crude) criterion for the validity of the hydrodynamic

<sup>1</sup>Check your favorite course on equilibrium statistical physics and thermostatics!

description of the gas motion is that the *Knudsen number*  $\text{Kn} \equiv \ell_{\text{mfp}}/L$  should be (much) smaller than unity. Do you see why? How can you interpret  $\text{Kn}^{-1}$ ?

Which other two (strong) inequalities should the three parameters of the “Boltzmann gas” obey? Are all three conditions compatible?

### 30. Relativistic kinetic theory

Educate yourself on the relativistic version of the Boltzmann kinetic equation, by reading Chapter I (& II?) of the book by de Groot, van Leeuwen & van Weert (link available on the lecture web page).